

## Large-Scale Fluid Transfer between Mantle and Crust during Supercontinent Amalgamation and Disruption

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**Abstract**—Supercontinents are a unique feature of the planet Earth. A brief review of supercontinents formed since the Archean shows that before the Eocambrian, supercontinents, notably Gondwana and Rodinia, amalgamated through high-temperature mobile belts, all of them containing ultrahigh-temperature granulite occurrences. During the final stage of the amalgamation, the lower continental crust was brought to magmatic temperature (from ~900 to more than 1000 °C) during a variable time span, from less than 10 Ma in the recent short-lived orogens to more than 150 Ma in the Eocambrian (Gondwana) or Neoproterozoic (Rodinia) long-lived orogens. Ultrahigh-temperature granulites worldwide contain the same types of fluid inclusions, namely, dense CO<sub>2</sub> and highly saline aqueous brines. The fluid amount in the peak metamorphic conditions is indicated by the amount of preserved fluid inclusions (especially CO<sub>2</sub>) and by the secondary effects caused by the fluids when they left the lower crust, including regional feldspathization, albitization or scapolitization, and formation of megashear zones, either oxidized (quartz–carbonate) or reduced (graphite veins). While some fluids may be locally derived either from mineral reactions or from inherited sediment waters, carbon isotope signature and petrographical arguments suggest that most fluids, both CO<sub>2</sub> and high-salinity brines, are derived from carbonatite melts resulting from partial melting of metasomatized mantle. Ultrahigh-temperature metamorphism is critical for supercontinent amalgamation, but the associated fluid causes instability and disruption shortly after amalgamation.

**Keywords:** supercontinents, ultrahigh temperature granulite, fluid inclusions, CO<sub>2</sub>, brines

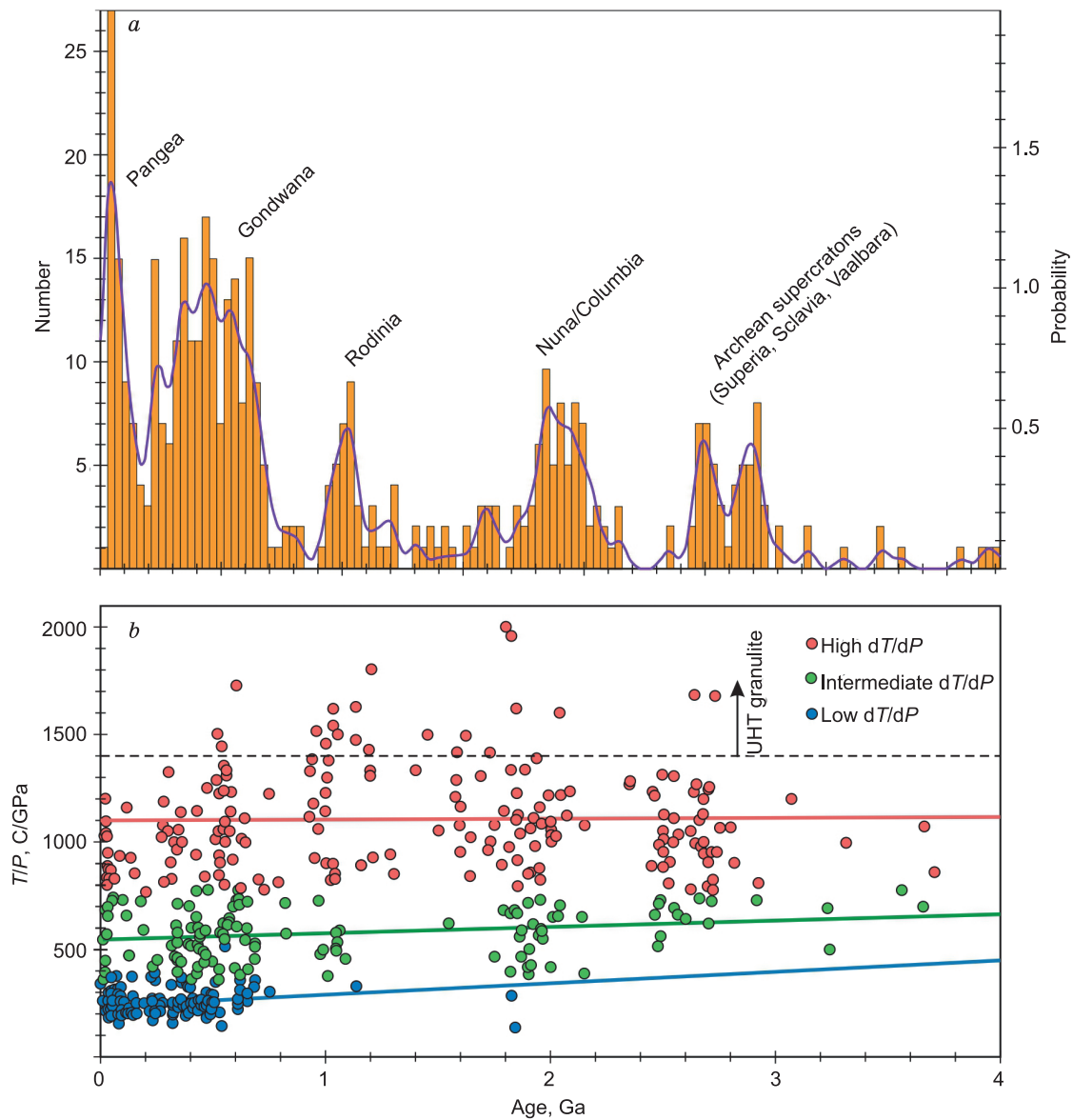
### INTRODUCTION

The Earth is the only planet in the Solar System to have continents, generated by the destruction of subducted oceanic crust since the early days of the Earth's formation. Throughout their history, continents remained mobile on the Earth's surface, periodically assembled in supercontinents, which after some time are fragmented into small pieces, en route towards a new supercontinent. No piece of continental crust has survived the late heavy meteorite bombardment, ca. 4 Ga, but the oldest gneiss found on Earth, generated very shortly after this event (Acasta gneiss, Canada, up to 4.28 Ga (Stern and Bleeker, 1998)) shows that continents already existed at this time. The oldest supercontinent, Vaalbara, is assumed to be about 3 to 4 Ga old (Cheney, 1996). The most recent supercontinent Pangea (about 0.3 Ga) is now in its final stage of disruption before the formation of a hypothetical new supercontinent in the future. The first over-

view of supercontinent formation was given by John Rogers in 1996: Ur, formed at ca. 3 Ga and accreted to most of East Antarctica in the middle Proterozoic to form East Gondwana; Arctica, a ca. 2.5–2 Ga continent that contained Archean terranes of the Canadian and Siberian shields and Greenland, and Atlantica, formed at ca. 2 Ga of cratons that now occur in west Africa and eastern South America. Arctica grew at ca. 1.5 Ga by accretion of most of East Antarctica and Baltica to form the continent of Nena (or Nuna). Collision of Nena, Ur, and Atlantica, plus minor plates, formed the supercontinent Rodinia at ca. 1 Ga. Rifting of Rodinia between 1 and 0.5 Ga formed three continents: East Gondwana, Atlantica (which became the nucleus for West Gondwana), and Laurasia (which contained North America, Greenland, Baltica, and Siberia). Gondwana formed at ca. 0.5 Ga by amalgamation of its eastern and western parts. Various plates accreted to Laurasia during the Paleozoic, and collision of Gondwana with Laurasia created Pangea at ca. 0.3 Ga. Although other names and assembly models have later been proposed, there is a general consensus on the limited periods of supercontinent occurrences (Fig. 1a).

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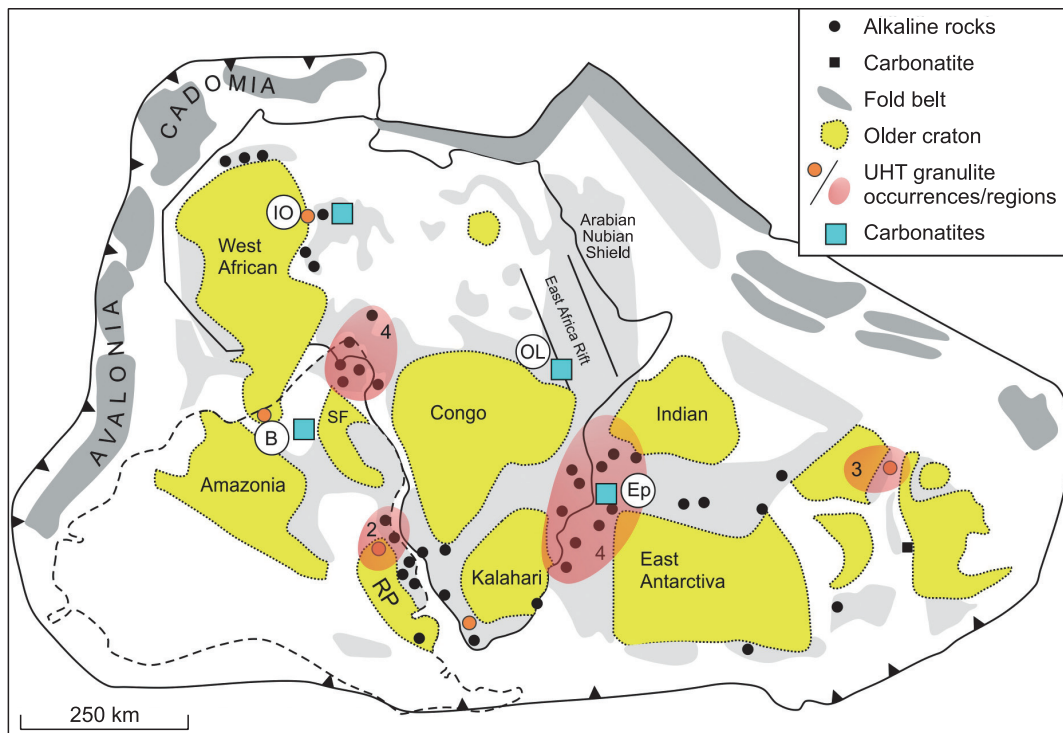
**Fig. 1.** *a*, Supercontinents throughout the Earth's history illustrated by metamorphic ages (histograms) recorded in 456 localities modified from (Brown and Johnson, 2018); *b*, metamorphic temperature gradients ( $dT/dP$ ) in these localities, subdivided into high, intermediate and low  $dT/dP$  groups modified from (Brown and Johnson, 2018). Colored solid lines: average values (low, intermediate and high  $dT/dP$  gradients, respectively); black dashed: separation between HT- and UHT granulites.

These include the mysterious Eoarchean (ca. >2.5 Ga) supercratons (Superia, Sclavia, Vaalbara), the better defined Nuna/Columbia (ca. 2 Ga), Rodinia (ca. 1.1 Ga), and finally the complicated Phanerozoic period, starting with the two semi-supercontinents (one in each hemisphere) of Laurasia and Gondwana, and ending with Pangea, presently in the process of disruption. Supercontinents older than Proterozoic, from Rodinia onwards, have been completely eroded, showing only their lower crustal roots. Disruption typically occurs by rifting along mobile belts, characterized by complicated relations between amalgamation and disruption. Laurasia and Gondwana started to break off immediately after the amalgamation that resulted in Pangea. Older super-

continents, on the other hand, had longer periods of existence, ca. 0.2 Ga for Rodinia and ca. 0.4 Ga for Nuna/Columbia.

#### SUPERCONTINENTS SEALED BY ULTRAHIGH TEMPERATURE METAMORPHIC EPISODES

All supercontinents are mainly made of metamorphic rocks, with no significant changes in high to intermediate thermal gradients since the Neoproterozoic (ca. 2.7 Ga) (Fig. 1b). Low thermal gradients are only abundant in recent orogens, but eclogites as old as 1.87 Ga have been reported

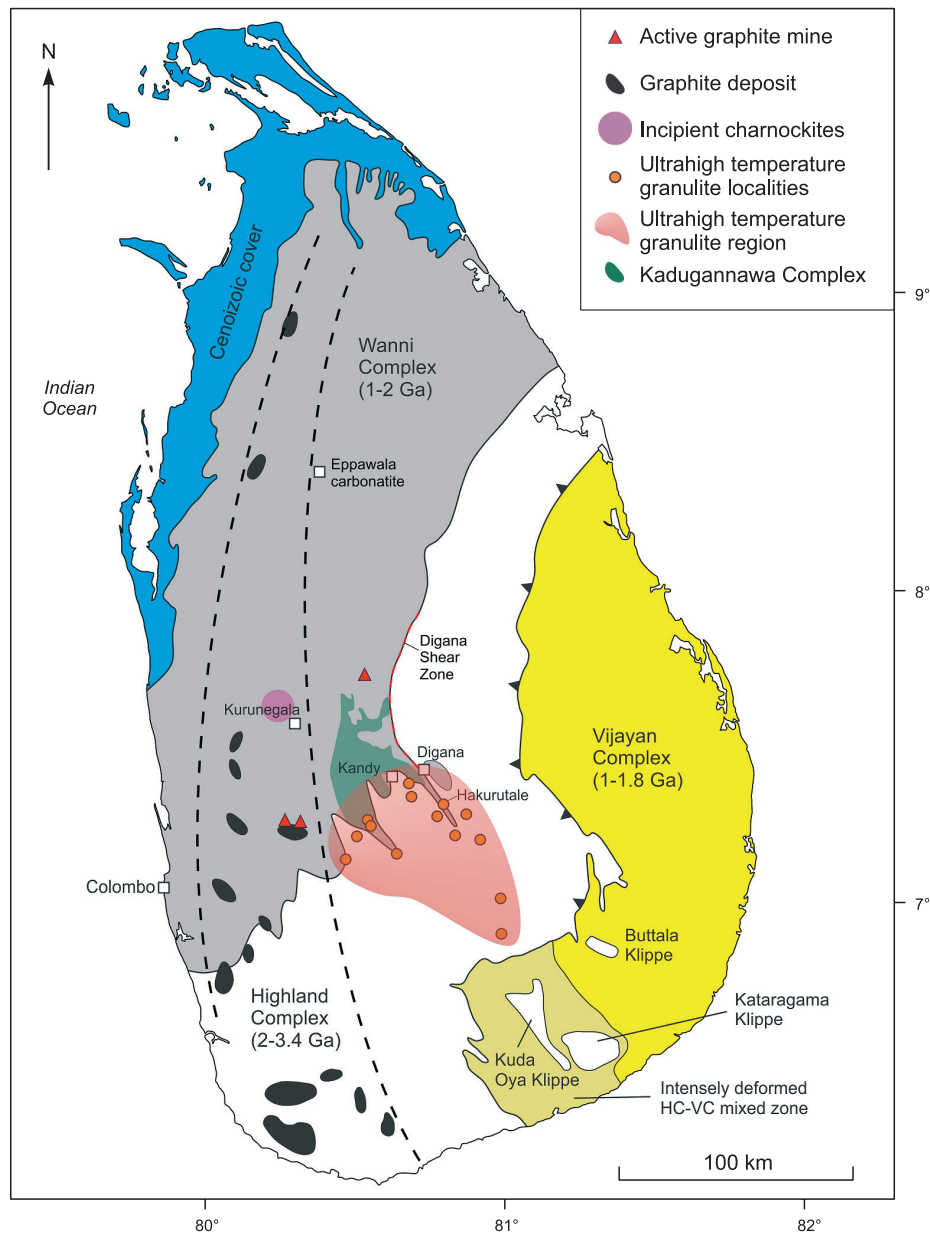


**Fig. 2.** Reconstruction of the Gondwana supercontinent at ca. 530–500 Ma modified from (Veevers, 2007) (yellow: older cratons, light-grey: Pan African mobile belts) except for Australia (uncertain age) and early Proterozoic spots along the margin of the West African craton (B, Bakhuis, Suriname; IO, In Ouzal, Sahara, Algeria.), UHT occurrences (after Kelsey and Hand, 2015) occur in Pan-African mobile belts connecting African cratons: Congo, India, East Antarctica (Madagascar, Sri-Lanka and Southern India, Antarctica), Central Africa and South America, Amazonia and West African craton. Ep, Eppawala, Sri-Lanka; OL, Oldyono Lengai, Tanzania).

to occur in the Kola Peninsula, Russia (Imayama et al., 2018). The regression line of low to intermediate  $dT/dP$  gradients shows indeed some lowering after ca. 0.8 Ga, but an even greater lowering also occurs with Archean rocks. Hence, a secular change in the thermal state of the crust cannot be the prime cause of the abundance of high-pressure rocks in recent orogens. It is more likely related to the level of erosion: after having been brought to great depth in collisional orogens, high-pressure rocks reached the greatest altitudes during post-collisional extension. Temperature regimes have not changed significantly since the end of Archean for the majority of lower-crustal rocks, which are assumed to be granulites (Rudnick and Fountain, 1995). Normal, high-temperature granulites constitute the greatest part of all cratons and correspond to high  $dT/dP$  metamorphic gradients, from ca. 800 °C/GPa, to more than 1200 °C/GPa. Extreme gradients, taken somewhat arbitrarily at  $dT/dP > 1400$  °C/GPa (Fig. 1b), characterize ultrahigh temperature (UHT) granulites, in which the metamorphic temperature exceeds 900 to 1000 °C. This temperature range covers the magmatic temperature range, at least for felsic rocks (charnockites). In these rocks, the difference between metamorphic and magmatic is non-existent; a magma that is emplaced at this temperature will show a metamorphic overprint after crystallization (Touret and Huizenga, 2012). The initial discovery of UHT granulites in the Napier Complex,

Antarctica, is relatively recent (Ellis, 1980). Currently, these spectacular rocks have only been found in the three most recent supercontinents (Columbia, Rodinia, and Gondwana), with a majority of documented occurrences in Rodinia and Gondwana (Kelsey and Hand, 2015).

For both Columbia and Rodinia, fragments of the supercontinents are so scattered that no clear relationships can be seen between UHT occurrences and supercontinent amalgamation. However, UHT occurrences in Southern Norway, either near the Rogaland anorthosite complex (Drüppel et al., 2015) or in the Bamble sector (Kihle and Bucher-Nurminen, 1992) are dated at ca. 1 Ga, which is close to the final stage of Rodinia amalgamation. The situation is more apparent for Gondwana. Figure 2 shows that most UHT occurrences in this supercontinent, including the type localities in Central Sri-Lanka and Antarctica, are located in the Pan-African mobile belt which did connect the Archean cratons during the final stage of Gondwana's assembly. They occur near intra-continental boundaries, the best places for finding traces of mantle plumes (Burov et al., 2007). The time of formation of this special case of extreme metamorphism is best established by single crystal dating of accessory minerals, notably zircon and monazite, e.g. SIMS or LA-ICP-MS (Kelsey and Hand, 2015). In Central Gondwana, UHT metamorphism starts in the Eocambrian (Pan-African orogeny, 0.6–0.5 Ga). Exceptions are two isolated spots



**Fig. 3.** The geology of Sri-Lanka (modified from (Touret et al., 2019b)) showing occurrences of graphite, key localities for incipient charnockites (Kurunegala), UHT granulites including Hakurutale (Bolder-Schrijver et al., 2000), the Digana Shear Zone (Binu-Lal et al., 2003), and the Eppawala carbonatite (Manthilake et al., 2008).

on the edge of the West African craton, Bakhuis, Suriname (Kroonenberg et al., 2016), or In Ouzal, Sahara, Algeria, (Bendaoud et al., 2004) (Fig. 2). Both are much older (early Proterozoic), possibly inherited from Columbia (Santosh et al., 2006). Note, however, that for these occurrences the regional geologic context is less known than in Gondwana, either because of political problems (Sahara) or because of limited exposure (Suriname). Despite these limitations, it does not change the main conclusion that UHT occurrences seal the final stage of the supercontinent amalgamation (Santosh et al., 2011).

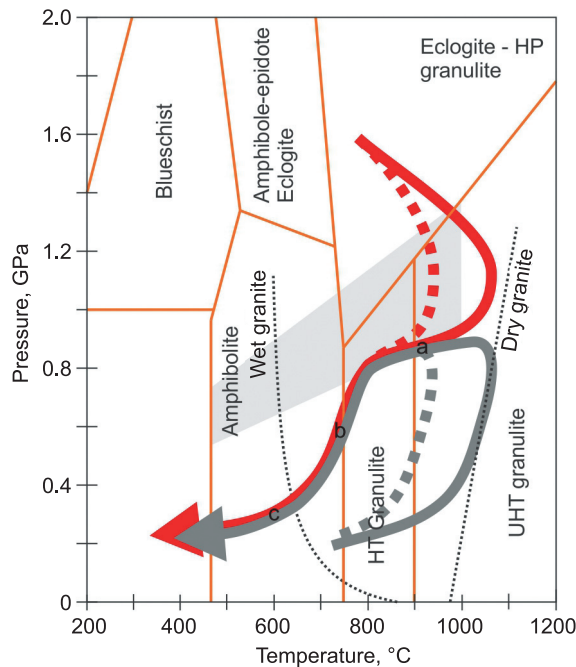
Figure 3 illustrates in more detail the mode of occurrence of UHT granulites in Central Sri-Lanka. They are found in a rather wide area within the high-grade Highland Complex, flanked on both sides by the lower grade Wannu Complex, to the West, and the Vijayan Complex to the East. The age of the protoliths is early Proterozoic to Archean for the Highland Complex and somewhat younger (1 to 1.8 Ga) for the two lower grade complexes. Like other UHT occurrences elsewhere, they form small spots, a few hundred of meters to a couple of kilometers across, dispersed within a rather well defined zone. As typical mineral assemblages (sapphi-

rine, osumilite) require high Mg–Al lithologies (e.g., Kelsey, 2008) it is highly probable that the actual number of UHT occurrences is higher than reported, but not enough to make continuous, kilometer-size outcrops. The isolated UHT domains typically grade into normal high-temperature granulites, equilibrated at peak temperature of about 850 °C.

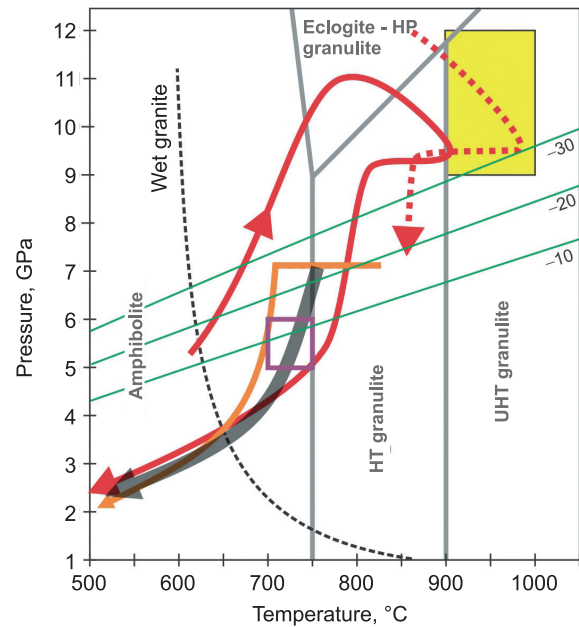
### *P–T–T* CONDITIONS OF UHT GRANULITE METAMORPHISM

With their exotic mineralogy and spectacular reaction textures, UHT granulites are among the best rock-types to perform geothermobarometry. Early temperature determinations (in the 2000’s) were based on element distribution (mainly Fe–Mg) in coexisting mineral pairs. More recent studies use complicated pseudo-sections for artificially defined chemical systems such as Fe–Mg–Al–Si (FMAS), later extended to include H<sub>2</sub>O (H) and potassium (K), and more recently Fe and Ti oxides (KFMASHTO) (details discussed in Kelsey and Hand (2015)). Pseudo-sections are calculated using various algorithms (Thermocalc, Perple\_X, Theriac Domino), making use of an internally consistent set of thermodynamic data. Results are illustrated by beautiful, highly complicated and spectacular phase diagrams (see e.g. Fig. 6 in Kelsey and Hand (2015)). Pseudo-sections seem to be a must for any modern paper in metamorphic petrology, but we contend that it is not easy to interpret them in terms of deriving actual *P–T* paths for at least two reasons: (1) the

artificial system will never reproduce completely the real composition; (2) the fluid phase is assumed to be either absent, or grossly simplified to be pure H<sub>2</sub>O whereas fluid inclusion data (or simple common sense) indicate the presence of concentrated NaCl-bearing aqueous solutions (see below). Our view is that more work is needed before pseudo-sections can fully represent the reality. Pseudo-sections are, however, helpful to reduce earlier estimates of peak temperatures by as much as 150 °C in the case of In Ouzzal (Doukkari et al., 2018). However, it is our opinion that, the general shape of UHT granulite *P–T* path is best approached by conventional thermo-barometry. The prograde path remains a challenge (Fig. 4): it can either be anticlockwise, characterized by a simultaneous increase in pressure and temperature to reach peak conditions, or clockwise when UHT granulites are formed from high-pressure, subduction-related rocks. In all cases, peak metamorphism corresponds to a sharp temperature increase (100 to 200 °C) (Fig. 4), suggesting an additional heat source at depth. The retrograde metamorphic evolution is marked by an initial episode of isobaric cooling followed by decompression. According to S. Harley (2008): “Many UHT terranes have evolved from peak *P–T* conditions of 8–11 kbar and 900–1030 °C to lower pressure conditions of 8 to 6 kbar whilst still at temperature in the range of 950 to 800 °C”. The



**Fig. 4.** Model *P–T* paths (clockwise and anticlockwise) for HT- and UHP/UHT granulites (dashed line: HT granulite, solid line: UHP/UHT granulite). In all cases, the post-metamorphic evolution is the same: initial isobaric cooling (a) followed by rapid decompression (b), and finally slow uplift towards the surface (c). Metamorphic facies are after (Brown, 2007).



**Fig. 5.** Pressure-temperature paths constructed for (1) the Highland Complex (red arrow) (Dharmapria et al., 2017); (2) UHT granulites from the Highland Complex (red dashed arrow, Dharmapria et al., 2015); (3) granulites from the Wannu Complex (orange arrow, Raase and Schenk, 1994; Touret et al., 2019b). The yellow box indicates peak *P–T* conditions for the UHT granulites in Highland Complex determined by Osanai et al. (2016). The purple box indicates the *P–T* conditions of charnockitization (5–6 kbar, 700–750 °C, Perchuk et al., 2000) at Kurunegala. The dark grey arrow indicates the *P–T* conditions of graphite vein formation (Touret et al., 2019b). Metamorphic facies are after Brown (2007).

rapid decompression is marked by partial destabilization of some minerals, notably garnet, to give typical symplectic decompression assemblage of orthopyroxene and plagioclase. This stage lasts until a temperature of ca. 600–500 °C, for a pressure drop of 2 to 3 kbar. It implies that rocks are at a temperature of ca. 500 °C at less than 10 km depth. The surface is attained by slow cooling during progressive erosion of the above lying exposed crustal levels. Figure 5 illustrates this type of  $P$ – $T$  path, documented for the UHT granulites of Central Sri-Lanka (Touret et al., 2019b), which can be taken as typical example for UHT granulites worldwide.

In contrast to the relatively consistent  $P$ – $T$  paths, the duration of UHT metamorphic episodes is rather variable. Based on single-zircon dating, Harley (2016) finds less than 15 Ma for the recent granulites from Hokkaido (Japan) or Seram (Indonesia) to more than 100 Ma in the long-lived and slowly-cooled Napier complex (Antarctica), the most classic example of a UHT terrane. Comparable values are indicated by Kelsey and Hand (2015): ca. 15 million years for Southern Madagascar and Southern India to more than 150 million years for the Eastern Ghats Province in India. Values in other regions vary regularly between these two extremes. These observations have led Kelsey and Hand (2015) to distinguish short (< 5 to 40 Ma) and long (> 40 Ma) timescale categories for UHT metamorphism. The reasons of this discrepancy are not clearly understood. Some authors believe that the cause for extreme UHT conditions are mainly crustal, caused by elevated heat crustal production and slow erosion rates (e.g. Clark et al., 2011, 2015). In India, the duration decreases progressively between the two regional metamorphic episodes, Archean or early Proterozoic in the north (ca. 2.4 Ga), Eocambrian (Pan-African) in the south. UHT metamorphism lasts for more than 150 Ma in the north (Eastern Ghats Province, which includes in its southern part the famous outcrops of Shevaroy or Nilgiri), 20 Ma in the Central Indian Tectonic Zone, which marks the limit between the two domains, and less than 10 Ma in the Pan-African rocks of Kerala and Sri-Lanka. It is hard to see any drastic changes in the lithology or erosion rates between these different domains, which in the field appear to be continuous. It is difficult to understand how crustal processes, e.g. rates of erosion or amount of heat-producing minerals could explain these differences in the duration of UHT-metamorphism. A more logical explanation for this is the presence of a mantle plume head, i.e. magma intrusions at the base of the crust or at the crust-mantle interface during the whole period of UHT metamorphism. This plume head has a different duration in short-versus long-lived orogens, but in all cases, the disruption of the supercontinent starts very shortly after the end of the plume activity.

## FLUIDS IN ULTRAHIGH TEMPERATURE GRANULITES

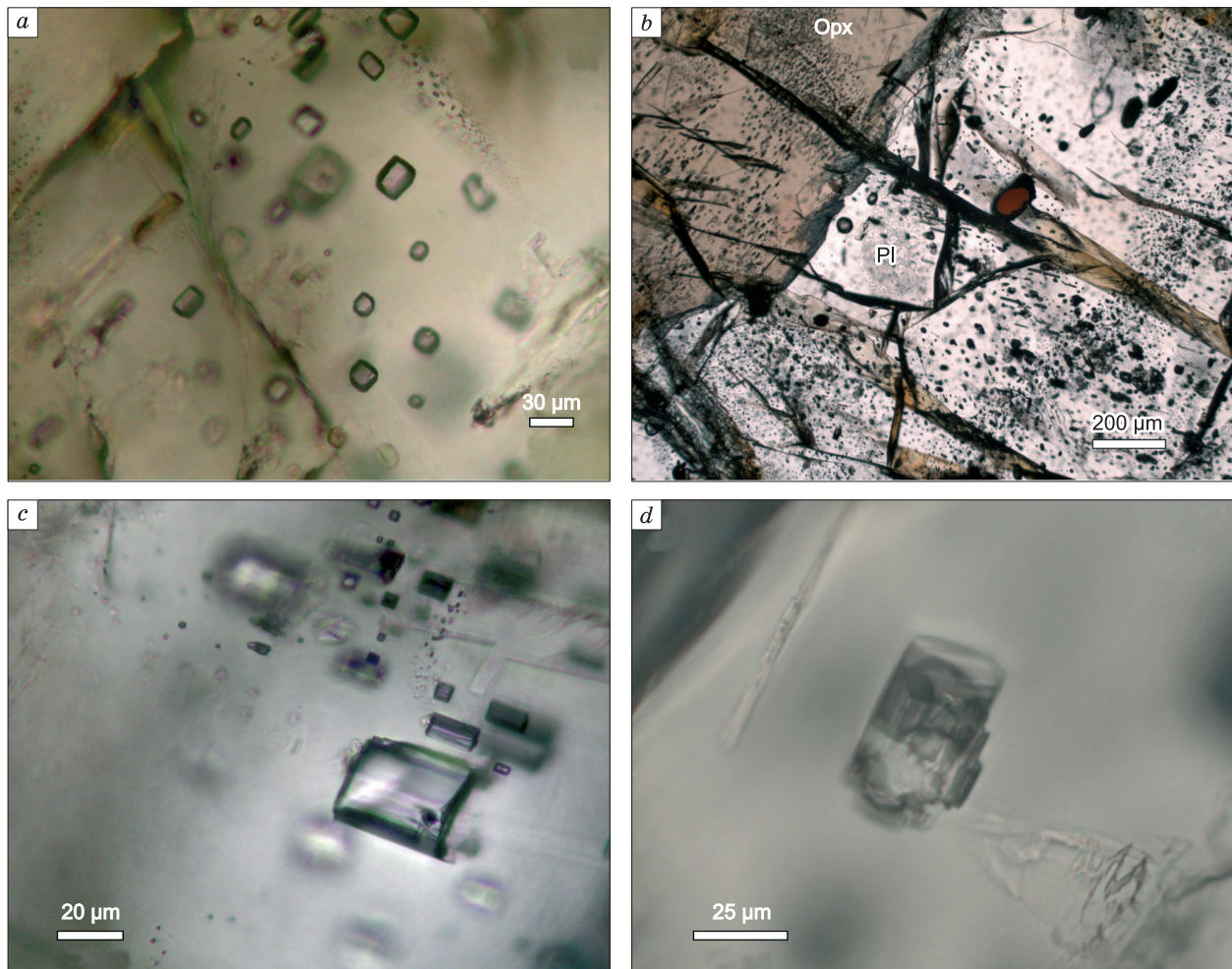
Virtually all granulites worldwide contain the same types of fluid inclusions, namely pure CO<sub>2</sub> of variable density and

highly saline aqueous solutions (brines) (e.g. Touret and Huizenga, 2011; Harlov, 2012). Ultrahigh temperature granulites are no exception, but they differ from normal high-temperature granulites in several ways.

**Abundance and density of CO<sub>2</sub> fluids.** Compared to other granulites, CO<sub>2</sub> inclusions in UHT granulites show three specific characteristics:

(1) Number and size of inclusions. Most rock-forming minerals (first of all orthopyroxene, plagioclase and garnet) contain many large (about 30–50 microns in size) and well preserved primary CO<sub>2</sub> inclusions, most of them with a nice negative-crystal shape. Typical UHT minerals (sapphirine, osumilite) are best found in Mg-Al-rich, silica-deficient rocks. Hence quartz is relatively rare, but when occurring it contains often very few or no inclusions at all, in sharp contrast to what is observed in most other rock types. This is probably due to the easy recrystallization of quartz at high temperature, which tends to wipe out all inclusions. Early publications on UHT granulites from former Gondwana have all insisted on the number of well-preserved, high-density and primary CO<sub>2</sub> inclusions: Napier Complex, Antarctica (Tsunogae et al., 2002), Shevaroy Hills, Southern India (Santosh and Tsunogae, 2003), Eastern Ghats Belt, India (Santosh et al., 2004), Sri-Lanka (Bolder-Schrijver et al., 2000). The same can be said for older UHT occurrences on the margin of the African craton (Fig. 3): In Ouzal, Sahara, Algeria (Cuney et al., 2007) and the Bakhuis, Suriname (this study). This list is not limited. In fact, we do not know any example of UHT granulites, which do not contain this type of primary, high-density pure CO<sub>2</sub> fluid inclusions. One example is given in Fig. 6, UHT granulites in sapphirine-bearing granulites from Udadigana, Harakutale, Central Sri-Lanka (Bolder-Schrijver et al., 2000). A photo can only show inclusions close to the surface of the section, but these are equally dispersed within the whole mass of the rock. A very conservative estimate of their number would be several 100's per mm<sup>3</sup>, corresponding to 5–10 vol.%. As most inclusions contain CO<sub>2</sub> of a density approaching (or exceeding) 1 g/cm<sup>3</sup> (see below), it indicates that the quantity of CO<sub>2</sub> still contained in the rock is of the order of 1 to 3 wt.%. The abundance of CO<sub>2</sub> at peak metamorphic conditions is further indicated by the high CO<sub>2</sub> content in channel minerals, first of all cordierite: 1.05 wt.% in the Napier Complex (S. Harley in Santosh et al. (2004)), up to ca. 3 wt.% in a metapelite from Bakhuis, Suriname, the highest level known from nature (De Roeve et al., 2016).

(2) Dense and super dense CO<sub>2</sub> fluid: The CO<sub>2</sub> density varies from sample to sample, but it is generally high to very high. In garnets from Salem (Shevaroy Hills, Southern India) UHT granulites, Santosh et al. (2004) found CO<sub>2</sub> fluid inclusions which homogenize within the narrow temperature range of  $-52.3 \pm 2.4$  °C, corresponding to densities of 1.17–1.16 g/cm<sup>3</sup>. These were at this time considered to be by far the highest density CO<sub>2</sub> yet reported from continental crust. However, even greater densities have been found in real super dense CO<sub>2</sub>, namely a fluid which homogenize be-

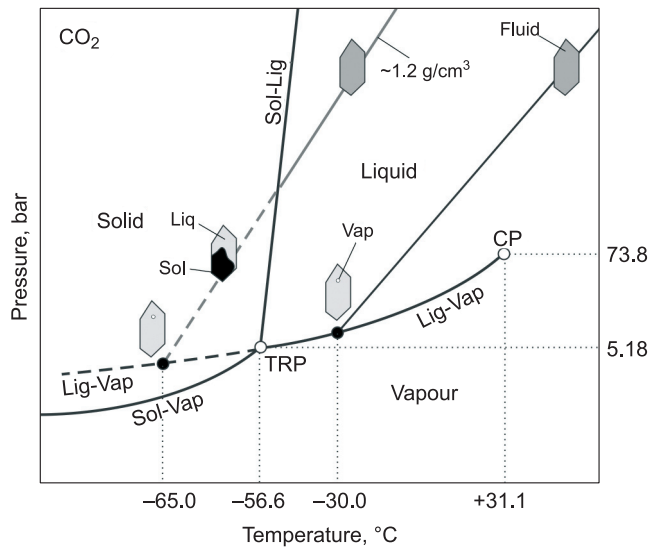


**Fig. 6.** CO<sub>2</sub> fluid inclusions in UHT sapphirine-bearing granulite from Harukutale, Central Sri-Lanka (Bolder-Schrijver et al., 2000). *a*, Group of primary CO<sub>2</sub> inclusions in plagioclase; *b*, illustration of the large fluid inclusion abundance (black dots) in plagioclase (Pl) and orthopyroxene (Opx); *c*, detail of primary, negative crystal shaped CO<sub>2</sub> fluid inclusions in plagioclase; *d*, dark CO<sub>2</sub> fluid inclusion containing a crystal of Mg-carbonate (white).

low CO<sub>2</sub> triple point (−56.6 °C), along the metastable extension of the liquid-vapor curve (Fig. 7). These remarkable fluids (Van den Kerkhof and Thiéry, 2001) are not restricted to UHT granulites. They can occur through late volume decrease of the inclusion cavity during post-metamorphic isobaric cooling (see discussion in Touret, 2001). However, in the case of UHT granulites, the primary character of the inclusions, the lack of any sign of post-trapping perturbation and the fact that fluid isochores match *P–T* conditions of mineral equilibration indicate with a good degree of confidence that these dense and super dense inclusions have really preserved fluids, which were present in the rock system at peak metamorphic conditions.

Minimum values measured so far (−65 °C) is in fluid inclusions in quartz in Bakhuis, Suriname, UHT granulites. It indicates a CO<sub>2</sub> density of ca. 1.2 g/cm<sup>3</sup> corresponding to a pressure of more than 10 kbar for a temperature of 1000 °C. The very good preservation of these primary inclusions, de-

spite the very high *P–T* metamorphic conditions, is best explained by the fact that, during the early stage of the retrogression, the pseudo-isobaric *P–T* path is roughly parallel to the high-density CO<sub>2</sub> isochores (Fig. 7). Starting above 1000 °C, this *P–T* trajectory typically continues to about 500 °C, at which the internal CO<sub>2</sub> pressure in the inclusion is of the order of ca. 5 kbar. This is followed by decompression from ca. 5 kbar to less than ca. 2 kbar occurs in less than 100 °C, resulting in the large decompression textures around garnet, found in many UHT granulites. The reason why CO<sub>2</sub> inclusions have better resisted to this decompression than many rock-forming minerals is not well understood. In normal, quartz-bearing granulites, such a decompression leads to prominent sets of secondary inclusion trails, progressively grading into open microfractures. An example is given in Fig. 8*a*, an incipient charnockite from Kurunegala, Sri-Lanka (Fig. 3) (Perchuk et al., 2000). In this rock, some orthopyroxene crystals grow inside a circular

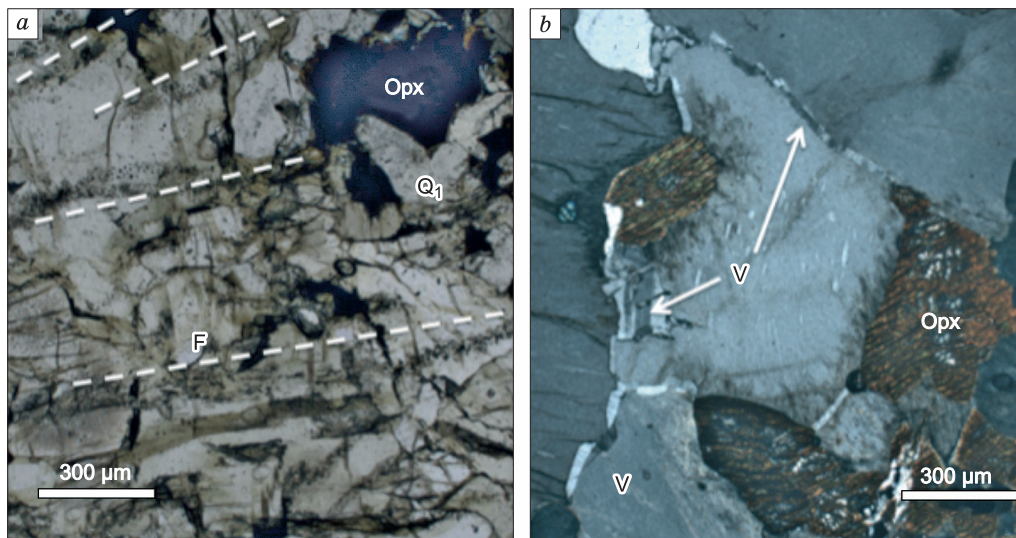


**Fig. 7.** CO<sub>2</sub> pressure-temperature diagram showing isochores for a dense (black solid line, Th is ca. -30 °C) and a super dense (dark-grey solid and dashed line, metastable Th is ca. -65 °C) CO<sub>2</sub> fluid inclusions. Th: temperature of homogenization into the liquid phase; CP, Critical point for pure CO<sub>2</sub> (+31 °C); TRP, triple point of pure CO<sub>2</sub> (56.6 °C).

cavity mantled by idiomorphic quartz crystals. Invariably one crystal, much longer than the others, displays a typical “Tessiner” habitus (Rykart, 1995), indicative of a CO<sub>2</sub>-rich fluid environment (Mullis, 1996). Many remnants of CO<sub>2</sub> inclusions are found in this quartz, showing that, at peak metamorphic conditions, free CO<sub>2</sub> fluids did exist in the rock, able to maintain open cavities at a crustal depth of at least 60 to 70 km. During retrogression, these fluids leave the rock system through a set of parallel healed fractures

(trails of secondary inclusions, Fig. 8a), progressively grading into open microfractures. The presence of these microfractures may be numerous, giving sometimes the impression of a secondary cleavage (unpublished observations of the senior author in the Bamble province of Southern Norway). Minerals like orthopyroxene or feldspars are less sensitive than quartz to secondary recrystallization, possibly helped by the equilibrium, the negative-crystal shape of the inclusions, and the non-reactive character of the CO<sub>2</sub> fluid.

(3) Carbonate inclusions in dense CO<sub>2</sub> fluids: As briefly mentioned earlier, many CO<sub>2</sub> inclusions contain solid carbonate, either a single, more or less cubic-shaped crystal, or an aggregate of several idiomorphic microcrystal. SEM images of opened inclusions indicate a Mg-rich composition of the carbonate (dolomite or magnesite) (Bolder-Schrijver et al., 2000). The volume ratio between carbonate and CO<sub>2</sub> and the composition of the carbonate are constant in all inclusions and not related to the inclusion shape or nature of the mineral host. In contrast to the carbonate globules mentioned earlier, these carbonates cannot be issued from a reaction between CO<sub>2</sub> and brines fluids (which would have consumed all CO<sub>2</sub>) or from a reaction between the CO<sub>2</sub> fluids and the mineral host (which would have produced carbonates of different size and composition, depending on the Mg–Ca composition of the mineral host). Homogenization temperature of the CO<sub>2</sub> in carbonate-bearing inclusions is about 20 °C higher than in carbonate-free inclusions. Consequently, the CO<sub>2</sub> isochore of the carbonate-bearing inclusions passes significantly below the *P–T* box of peak metamorphic conditions, but it approaches those *P–T* conditions if the molar volume of the solid carbonate is reintegrated into the total molar volume of the fluid-solid system (see discussion in Bolder-Schrijver et al., 2000). For these rea-



**Fig. 8.** Contrasting mechanisms of fluid migration in incipient charnockites from South India and Sri-Lanka. *a*, Kottavattam, Kerala (Touret et al., 2019a). Orthopyroxene (Opx) is growing in a cavity (geode) mantled by idiomorphic quartz crystals with a Tessiner habitus (Q<sub>1</sub>). Post-metamorphic fluids are trapped as secondary fluid inclusions along healed microfractures (F) *b*, Kurunegala, Sri-Lanka (Perchuk et al., 2000). V = K-feldspar microveins along intergrain boundaries between quartz and perthite phenocrysts.



sons, we believe that the incoming CO<sub>2</sub> in UHT granulites at Harakutale at peak metamorphic conditions was not pure CO<sub>2</sub>, but a mixture of CO<sub>2</sub> and carbonate. We have hypothesized that the source could be a carbonate melt, partly destabilized when entering the lower crust (Touret and Huizenga, 2012b).

**Brine fluids.** Besides CO<sub>2</sub>, UHT granulites contain also traces of high salinity aqueous fluids (brines), like all other granulites (Touret and Huizenga, 2011). These fluid inclusions are less abundant and smaller than CO<sub>2</sub> inclusions, and, therefore, much more difficult to be studied. A liquid phase is commonly absent, inclusions are made of a small, irregular cavity squeezed around one or several solids, the larger being invariably isotropic halite (referred to as collapsed inclusions, (Touret, 2001)). Isochores for brine fluid inclusions are invariably steeper than those for CO<sub>2</sub>, i.e. the fluid pressure in brine inclusions is less than the lithostatic pressure during the early stage of the retrogression, unable to remain unchanged at these high temperature. But the fact that they were already NaCl saturated implies a very high salinity, at least 50 wt.%. Luckily, some fluid inclusions have escaped this transposition, one example being the Bakhuis UHT granulites (Surinam, South America), in which isolated brine inclusions occur within groups of more abundant CO<sub>2</sub> inclusion (Fig. 9a). With a few exceptions, e.g. in In Ouzzal UHT granulites (Cuney et al., 2007), immiscible brines and high-density CO<sub>2</sub> fluids occur in virtually all UHT granulites studied so far. Figure 9b shows another example in an igneous charnockite occurring in Southern Norway. This rock belongs to a granulite “island”, north of the regional amphibolite-granulite isograd (Nijland et al., 1998). The high metamorphic temperature of this rock

(>900 °C) is indicated by feldspar mesoperthites. Similar rocks are found in most UHT granulite occurrences, indicating that the very high temperature reached is caused by magmatic intrusions during the peak of metamorphism.

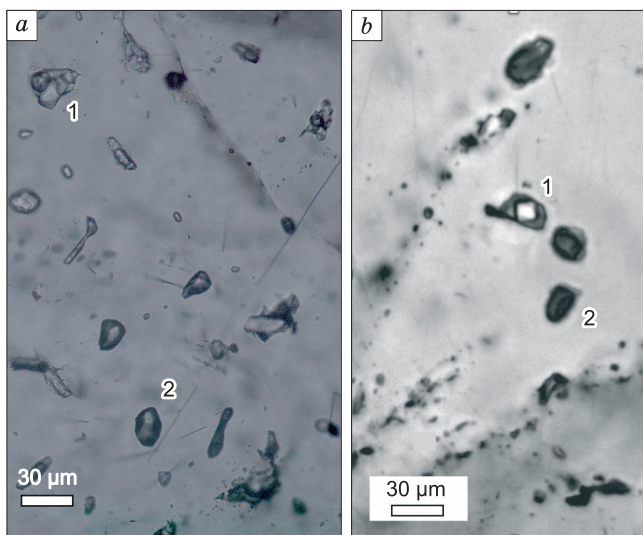
Brine inclusions are less common than CO<sub>2</sub> inclusions, especially in UHT granulites. However, their role in rock transformation is significant due to their mobility and capacity of element transport. They are instrumental for the formation of the so-called “incipient charnockites”, namely the local (cm to m scale) transformation of gneiss into igneous-looking charnockites (Touret et al., 2019a), as will be discussed in more details in the following section.

In conclusion, fluid inclusion studies demonstrate without ambiguity that free fluids did exist in the rock system at peak metamorphic conditions, in the form of dominant dense to super dense CO<sub>2</sub>, accompanied by immiscible brines. Then the next obvious questions arise: how much fluids were present and where do they come from?

## HOW MUCH FLUID: INDIRECT EVIDENCE

Well-preserved fluid inclusions only provide a minimum estimate of the total fluid quantities present in the rock system at the time of the inclusion formation. As discussed above, this minimum is far to be negligible, especially for CO<sub>2</sub>. On the other hand, another indication is given by the extraordinary variety and extent of secondary effects caused by metamorphic fluids at the onset of regression. These effects have been discussed elsewhere (Touret and Nijland, 2013; Newton et al., 2019; Touret et al., 2019a) and only a brief summary will be given here.

Fluids are introduced in the lower crust at peak UHT-conditions in the form of immiscible droplets of high-density CO<sub>2</sub> and high-salinity brines, as found e.g. in Bakhuis, Suriname (Fig. 9a). However, their reactivity and migration capacity are grossly different, highly reactive brines being much more mobile along intergrain boundaries. Besides the often quoted difference in wetting angle (Watson and Brennan, 1987; Harlov, 2012), a major cause for the difference in migration capacity is to be found in the systematic difference in fluid isochore slope between carbonic and aqueous systems (see discussion in (Touret, 2001)). Flat high-density CO<sub>2</sub> isochores are roughly parallel to the isobaric cooling *P-T* path, which occurs systematically after UHT-metamorphism. Internal fluid pressure in the fluid inclusion remains similar to the lithostatic pressure until the end of isobaric trajectory. Only during the subsequent sudden decompression, e.g. in Sri-Lanka at a temperature of about 700 °C (Touret et al., 2019b), CO<sub>2</sub> will migrate through a network of healed microfractures. Alternatively, aqueous fluid inclusions have a fluid pressure, which is less than the lithostatic pressure at the onset of regression, resulting in successive transpositions, allowing the fluid to reach the grain boundary of the host mineral.



**Fig. 9.** Examples of primary brine (1) and dense to high-density CO<sub>2</sub> (2) fluid inclusions demonstrating immiscibility in UHT granulites at peak metamorphic conditions: *a*, Bakhuis, Suriname; *b*, Igneous charnockite, Bjornasvatn, Southern Norway. In the lower part of the section, trails of smaller secondary inclusions correspond to the post-metamorphic escape of the primary fluids.

The result is that both fluid systems behave independently, not interacting with each other until the final stage of regression at ca. 300–400 °C, at which brine fluids can react with CO<sub>2</sub> to give late carbonates (seen systematically in all granulites as micro-inclusions, notably in feldspars). Free CO<sub>2</sub> remains exclusively within the granulite domain. Its geochemical influence is minor, restricted to the stabilization of the characteristic granulite anhydrous mineral assemblage (first of all, orthopyroxene). Alternatively, saline aqueous fluids circulate pervasively in and outside of the HT and UHT granulite areas with important geochemical consequences, especially within the igneous looking charnockites (Touret and Huizenga, 2012). In the granulite domain, brine streaming along intergrain boundaries results mainly in the formation of K-feldspar microveins along quartz-plagioclase or plagioclase-plagioclase grain boundaries (e.g., Harlov and Wirth, 2000). When they reach the outer limit of the granulite areas, they are able to cause the transformation of various lithologies (metapelites, amphibolites, and rocks belonging to the typical Archean trondhjemite, tonalite, granodiorite (TTG) series) into coarse-grained, igneous looking charnockites. These incipient charnockites were first identified in the Archean Darwar craton of India (Pichamutu, 1960) (Fig. 3a) and later in the Neoproterozoic Province of South India (Kerala) and Sri Lanka. They occur at close distance (typically a few 100 m to km) of the regional granulite/amphibolite facies boundary in most HT/UHT granulite terranes. The fact that some incipient charnockites occur along a network of brittle fractures led most authors to postulate that they were caused by fluid streaming (Friend, 1985), first believed to be CO<sub>2</sub> when it was thought that CO<sub>2</sub> was the only active granulite fluid (e.g. Santosh et al., 1990). However, a recent re-investigation of fluid inclusions in incipient charnockites from type localities in Southern India (Touret et al., 2019a) confirms without ambiguity that brines are the fluids responsible for incipient charnockitization; CO<sub>2</sub>, when present, is a consequence and not the cause. Besides brittle fractures, incipient charnockites may also occur as rounded patches, with textures showing evidence of melting (Perchuk et al., 2000, Touret and Huizenga, 2012a) (Fig. 3b). A detailed investigation of the conditions imposed by the brines (*P*, *T*, H<sub>2</sub>O activity) shows indeed that gneiss melting can occur under the influence of brine streaming, suggesting this mode of formation for a number of massive charnockites (Newton et al., 2019). Brines may focus along large shear-zones to invade the adjacent amphibolite domain, as seen in India north of the so-called Fermor line, the northern limit of Archean granulites. The occurrence in this domain of elongated granite bodies (including the famous Closepet granite) suggests a new way to look at the relations between charnockite and granite in the low to middle crust (Newton et al., 2019). The Closepet granite was formed at a lower pressure than the charnockite (4 to 8 kbar (Moyen et al., 2003)). At a relatively low pressure (ca. 4 kbar), CO<sub>2</sub> fluids are expelled out of the crystallizing magma in the form of veins (Battacharya et al., 2014). Biotite remains

stable and the final rock is a granite. The relationships between charnockite and granite appears then to be strictly fluid-dependent: high-salinity aqueous brines induce the metasomatic-melting effects, while CO<sub>2</sub>, if pervasively present, stabilizes the charnockite diagnostic minerals including orthopyroxene. The systematic occurrence of saline inclusions in potassic granites and syenites, unrelated to boiling except for very shallow intrusions (Cu-porphyrates (Roedder, 1984)) (e.g., Konnerup-Madsen, 1984), suggests that this process of fluid streaming through the crust might be important for the formation of granites in general.

In all granulite terranes, high-temperature transformations are continued at decreasing temperature by intense alkali metasomatic effects, which under various names (albitization, scapolitization, K-feldspathization) can cover region of subcontinental size, e.g. in the southern part of the Scandinavian shield (Sweden, Hoeve, 1978; or Norway, Engvik et al., 2014). This type of metasomatism, which has important metallogenic consequences (Fe-Ti ores in Norway, U in Sweden and elsewhere), is far more developed in the regional amphibolite-grade gneisses than in the granulite core. The source of brines is probably multiple, but in Southern Norway there is compelling evidence (Touret and Nijland, 2013; Engvik et al., 2014) that brine fluids were released during prograde metamorphism of pelitic sediments, including evaporites. These fluids were present almost everywhere at the onset of peak granulite metamorphism, closely associated with metasediments. Besides halides, sulfates (notably anhydrite) are common daughter minerals in inclusions, explaining the highly oxidized character of many granulitic areas (Harlov et al., 1997). It is not certain, however, that the example of Southern Norway is valid everywhere. The carbonatite connection for the origin of CO<sub>2</sub>, discussed below, offers another possibility, which needs to be further documented.

#### EFFECTS INDUCED BY CO<sub>2</sub> FLUIDS: QUARTZ-CARBONATE AND GRAPHITE MEGASHEAR ZONES

The widespread occurrence of late- to post-metamorphic Na-metasomatism indicates that most metamorphic Na-bearing aqueous fluids left the rock at the end of the metamorphic episode. However, a significant quantity of brine fluid may remain trapped along the intergrain boundaries, as indicated by high-conductivity layers at the base of most continental areas (Manning, 2018). The conservation of CO<sub>2</sub> is more problematic, except for the fluid remnants preserved in mineral inclusions. Many granulite terranes are either intersected or surrounded by prominent crustal-scale shear zones, sometimes up to ca. 100 km long with a width of ca. 1 km showing evidence of large scale, fluid-induced metasomatism. Some, containing mainly pure quartz, have been termed carbonated shear zones, displaying pervasive and vein-controlled carbonate metasomatism (Newton and Man-

ning, 2002). Examples include the late Proterozoic Valley of Tamil Nadu, India (Wickham et al., 1994), the Late Archean Chitradurga area of Karnataka, India (Chadwick et al., 1989) and the mid-Proterozoic Bamble Shear belt of Southern Norway (Dahlgren et al., 1993). On the contrary, other shear zones are highly reduced, with the best example being the vein-type graphite of Sri-Lanka, the world largest resources of highly crystalline graphite (Fig. 3). Graphite veins are also surrounded by potassic feldspar alteration zones of variable size (Touret et al., 2019b). Fluid inclusion studies and  $P$ - $T$ -oxygen fugacity interpretation of the conditions of formations of the different veins (see discussion in (Touret et al., 2019b)) demonstrate the involvement of both granulite fluids in the two types of shear zones redox settings: brines in the oxidized shear zones, and  $\text{CO}_2$  in a reduced setting. In the quartz-carbonate megashear zones, the high oxygen fugacity is imposed by the sulfate-bearing brines (Newton and Manning, 2004), whereas the reduced character of the graphite veins is caused by abundant sulfide minerals in the host-rocks (Katz, 1987). During the reduction of  $\text{CO}_2$ , the residual fluid becomes enriched in  $\text{H}_2\text{O}$  (e.g., Huizenga, 2011; Touret et al., 2019b), which is responsible for the alteration zones found on both sides of the graphite veins.

Megashear zones are the most spectacular effects induced by the release of lower crustal  $\text{CO}_2$  fluids at the end of the UHT-metamorphic episode. They are the best illustration of the fundamental difference in the mechanisms of migration of brines and  $\text{CO}_2$  fluids within the crust (Fig. 8). Successive episodes of brittle deformation must have caused micro seismicity, grading into more significant seismicity as indicated by the frequent occurrence of pseudotachylites in granulites (Clarke and Norman, 1993). Propagating through the entire crust, this mechanism is a very effective way to transfer deep  $\text{CO}_2$  towards the surface, as shown by the persistent  $\text{CO}_2$  emissions following major earthquakes (Girault et al., 2018). This effect is, however, limited in geological time. Alternatively, the common occurrence of high-conductivity layers at the base of continents indicates that a significant quantity of brine fluids, probably stored along mineral inter-grain boundaries, might remain within the lower crust at the end of the metamorphic episode (Manning, 2018).

#### **FLUID ORIGIN: THE CARBONATITE CONNECTION**

The dry character of granulite terranes has led to the widely accepted concept of fluid-absent (or vapor-absent) rocks, in which any free fluid, tacitly or explicitly assumed to be water, cannot exist (Thompson, 1983). Syn-metamorphic fluid is either dissolved in melts, or bound in mineral structures. This is indeed the case for pure water: at HT- or UHT metamorphic temperatures, any free water will immediately provoke partial melting and be dissolved in the melt. This mechanism is quite obvious in all high-grade rocks, but, contrary to common belief, melts do not leave the sys-

tem systematically. Many granulites are migmatites, in which both initial lithology and products of partial melting still coexist at the outcrop scale. Melts are only expelled towards more superficial levels if produced in sufficient quantity or in zones of special deformation. As the rate of melt production depends on the availability of  $\text{H}_2\text{O}$  (relatively little in the dry granulite terranes), granite melts are far more abundant in amphibolite-grade migmatites than in granulites. However, in these, partial melting is a mechanism able to generate both  $\text{CO}_2$ , when the protolith was carbon-bearing (e.g., metapelites), and brines, by consuming progressively the water released by crystallizing melts to build up hydrous minerals (desiccation effect (Markl and Bucher, 1998)). Both processes have been documented, but they are not likely to produce large fluid quantities. This is evident for brines, which, according to the desiccation mechanism, can only be minor remnants strictly confined to former granitic melts. For  $\text{CO}_2$ , burning of former organic matter, notably in metapelites, can provide important  $\text{CO}_2$  quantities. This is indeed observed in the Bamble province of Southern Norway, which contain a number of graphite-bearing paragneisses, but these contain also sulfides, leading to immediate reduction of high-temperature  $\text{CO}_2$  to form graphite during retrogression (Huizenga and Touret, 2012). Another possible source of carbon would be sedimentary carbonates, occurring in many granulite areas in the form of marbles or, in Bamble, in garnet-clinopyroxene bearing skarns. In Southern Norway, these skarns contain the major iron-ore deposits (Arendal region), but, interestingly, only brine inclusions have been found in these rocks;  $\text{CO}_2$  is very rare or completely absent (Touret, 1985). The fact that sedimentary carbon can only make a minor part of the high temperature  $\text{CO}_2$  found in HT and especially UHT granulite is further indicated by the isotopic signature of the inclusion fluid. Inclusions in felsic granulites from the Bamble area contain isotopically light carbon ( $\delta^{13}\text{C} = 24.2$  to  $14.1\%$ ), interpreted as a mixture between light organic and a variable proportion of heavier carbon, either a minor quantity of sedimentary carbonate or some mantle-derived  $\text{CO}_2$  (Pineau et al., 1981). The mantle signature is more clear for synmetamorphic carbonates found in partly amphibolized basic intrusions (hyperites), emplaced at peak metamorphic conditions ( $\delta^{13}\text{C} = 8.2 \pm 1\%$ ). These data refer to HT granulites, but UHT granulites show much more clearly a dominant, if not exclusive mantle signature, either from noble gases (mainly He) (Dunai and Touret, 1993) or from the carbon isotope composition of graphite in Sri-Lanka (Touret et al., 2019b).

Due to the lack of a suitable geochemical tracer, the origin of brines remains uncertain (e.g., Manning, 2018). Brines issued from metasediments, including former evaporites are widespread in Bamble area in and out of the regional granulite area (Touret, 1985). Derived from former sedimentary formations waters, these brines are probably responsible for regional albitization-scapolitization or for the formation of granulite islands, but we have no evidence that they could participate to an immiscible brine- $\text{CO}_2$  sys-

tem at UHT peak conditions. Alternatively, the association of CO<sub>2</sub> and carbonates, both with a mantle signature, indicates a clear relation with carbonatite melts. Such an origin fits also well with the rather limited size of the UHT granulite occurrences: carbonatite melts can be transported by magmas emplaced at the base of the crust, providing both the heat and fluids of UHT granulite metamorphism. This carbonatite connection is further supported by the many intrusions of carbonatites and associated rocks (nepheline-syenite and other types of alkaline intrusions) found in the vicinity of in former UHT areas. Figure 3 illustrates the great number of alkaline intrusions in Gondwana UHT occurrences. Typical carbonatites are less abundant, but some are found in all regions mentioned in this paper, e.g. Eppawala, Wann Complex, Sri-Lanka (Weerakoon et al., 2001; Mant-hilake et al., 2008), Tamil Nadu, India (Natarajana et al., 1994), In Ouzzal, Algeria (Ouzegane et al., 1988), and Bakhuis, Suriname (Kroonenberg et al., 2016). These intrusions are of different age: Paleoproterozoic in Tamil Nadu or along the margin of the West African craton (In Ouzzal and Bakhuis), Eocambrian (Pan-African) in Central Gondwana. They post-date the metamorphic episodes by a variable time span: tens of millions of years in most cases, up to several hundred million years when they are related to large-scale rifting in the consolidated cratons: e.g., the Fen carbonatite in Southern Norway (Permian Oslo graben) and active carbonatite volcanism in the East Africa Rift.

The carbonatite connection offers another possibility for the origin of brines in the UHT granulites. Salts are common constituents of carbonatites (e.g., Safonov et al., 2010) which contains inclusions strikingly similar to those found in UHT granulites (Guzmics et al., 2011). Rapidly grown fibrous diamonds contain highly saline fluid inclusions. These brine fluids were trapped at a depth of at least 200 km and derived from silicic and carbonatitic deep mantle melts (Weiss et al., 2015). Fluids issued from subducted, salt-bearing altered oceanic crust (Sharp and Barnes, 2004) cause mantle metasomatism, which is the source of carbonatite melts (Meen, 1987). Remnants of chloride-carbonate melts have also been found in diamonds from metasomatized mantle eclogites (Zedgenizov et al., 2018). Spectacular associations of carbonate and halite have been observed in Udachnaya-East kimberlite (Kamenetsky et al., 2012; Abersteiner et al., 2018) (Figs. 13–8, p. 821 in (Frezzotti and Ferrando, 2018)). Chlorides are found in many kimberlite worldwide but the Udachnaya kimberlite is unique by its extremely high chlorine concentration (up to 6 wt.%) and the presence of chloride-bearing segregations. It is, however, not clear in this case if the salt is magmatic or resulted from evaporite contamination during kimberlite eruption (Grishina et al., 2014a,b; 2018). However, besides this discussed example, there is ample evidence that, in general, carbonatites can provide a substantial amount of the salts found in UHT fluids, with significant geochemical consequences. Carbonatite melts are formed by partial melting of a mantle metasomatized by fluids issued from former sub-

duction. That these fluids may carry salts is not surprising. More surprising, though, is the fact that the mantle metasomatic source may have remained at the same place for millions of years, despite the complicated history of the overlying crust.

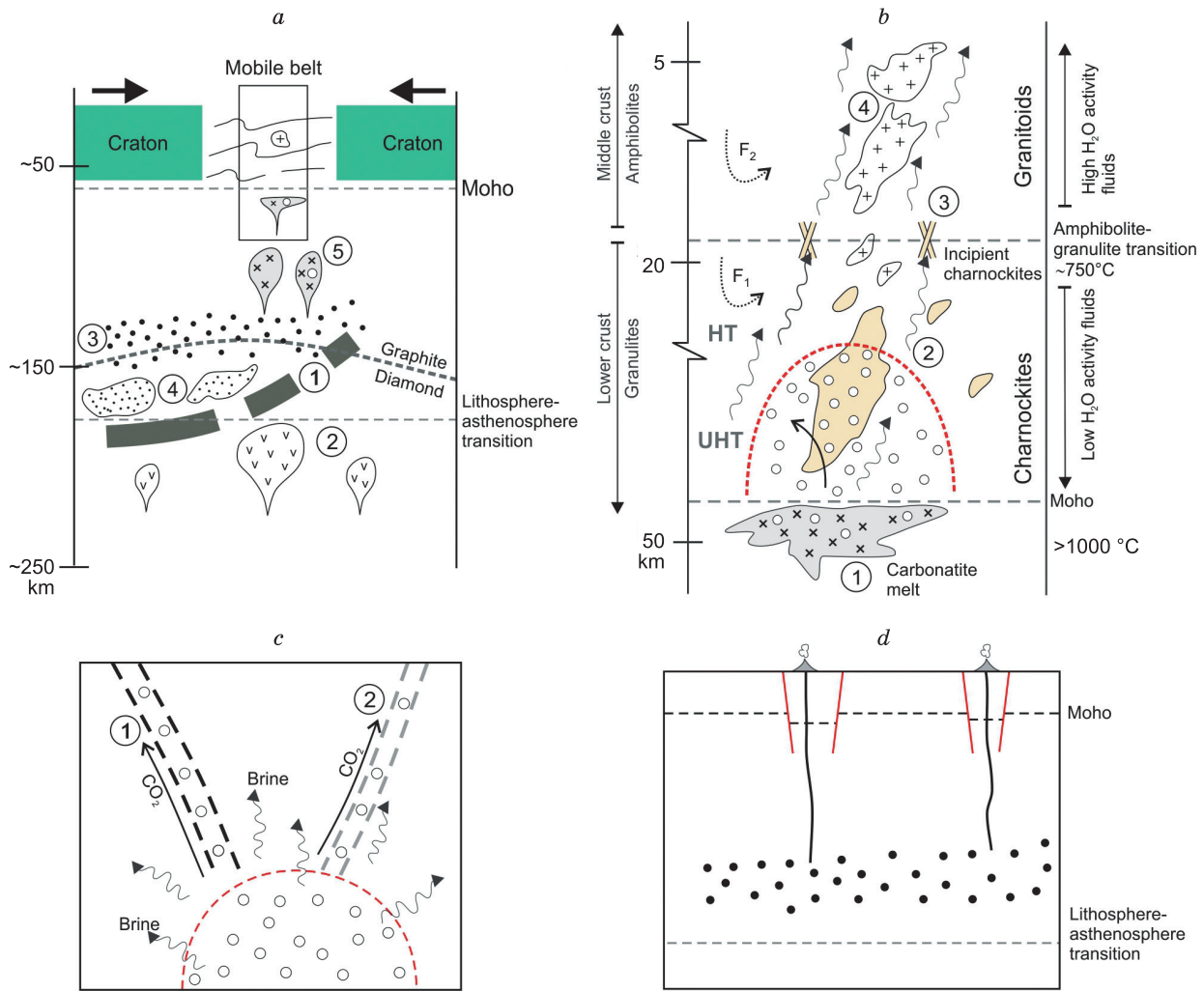
#### ULTRAHIGH TEMPERATURE AND SUPERCONTINENT AMALGAMATION: A SYNTHESIS

Petrological evidence and fluid remnants contain in mineral inclusions indicate that the lower continental crust was invaded by large quantities of mantle-derived CO<sub>2</sub> fluids during episodes of UHT metamorphism. This CO<sub>2</sub> might be accompanied by high-salinity aqueous brines, or mix with internally derived fluids to form an immiscible CO<sub>2</sub>-brine system. UHT occurrences were relatively limited in space, related to the proximity of synmetamorphic magmatic intrusions and responsible for the very high temperature attained during peak UHT metamorphism. UHT occurrences were instrumental for sealing the final stage of supercontinent amalgamation. However, the large fluid amount stored in the lower crust might have caused instability, preparing a disruption which has started immediately after amalgamation (Touret and Huizenga, 2012). Post-metamorphic evolution was determined by the physical-chemical properties of the migrating fluids: pervasive granulitization and alkali metasomatism for the mobile brines, brittle fracturing for the less mobile CO<sub>2</sub>. This type of evolution is best documented for the Gondwana supercontinent, but there are indications that it was also valid for older supercontinents including Rodinia and Columbia.

Figure 10 illustrates tentatively the details of this four-stage evolution, based on the Gondwana UHT occurrences (South India and Sri-Lanka). During final amalgamation of a supercontinent (Fig. 10a), two cratonic blocks are separated by a mobile belt, which seals the amalgamation. Ancient altered oceanic crust, subducted near the asthenosphere/lithosphere boundary, liberates fluids and melts which metasomatize the overlying mantle. Remnants of these fluids can be found in inclusions in diamonds, e.g. saline fluids issued from melted altered oceanic crust and/or sediments. During this process, various types of restite material can be formed, notably garnet-bearing eclogite and lherzolite (Simakov, 2003).

The amalgamation starts with an episode of UHT metamorphism in the lower crust of the mobile melt, triggered by a plume head rising from the deep mantle. It induces melting of the metasomatized mantle and the formation of carbonatite melts, immiscible in silicic melts and carried upwards in these melts, of average basaltic composition.

The extreme temperature of UHT metamorphism at the base of the crust (>1000 °C) is sufficient to maintain temperatures above minimum granite crustal melting in most of the continental crust, up to a depth of few kilometers



**Fig. 10.** Synthesis of UHT granulite and supercontinent amalgamation (see text for a detailed explanation). *a*, Supercontinent final amalgamation stage. 1, Subducted former altered oceanic crust; 2, Rising deep-mantle plumes; 3, Metasomatized mantle; 4, Restite from melted subducted crust (garnet-bearing eclogite and lherzolite); 5, Carbonatite melts and basaltic melts carrying immiscible carbonatite melts; *b*, detail of the insert (mobile belt) in *a*. 1, mantle intrusions carrying immiscible carbonatite melts. The dashed red line indicates the accumulation of dense/hyper dense CO<sub>2</sub> (open dots) in UHT granulite domains; 2, lower crustal charnockite (massive). 3, incipient charnockite; 4, middle crustal granites (for example Closepet granite, South India) including coarse-grained monzogranite and fine-grained subvolcanic granite. F1 and F2 (fine-dashed arrows): Locally derived fluids in the lower- and middle crust, respectively. Thick arrow: CO<sub>2</sub>, thin contorted arrow: brine; *c*, Elimination of CO<sub>2</sub> in lower crust UHT granulites during decompression and cooling. 1, graphite veins; 2, quartz-carbonate mega-shear zones; *d*, carbonatite eruptions in the stabilized craton triggered by extension.

(Fig. 10*b*). It will generate a variety of granitoid magmas, possibly mixed with differentiated melts issued from the mantle. The example of Southern India (Karnataka, Madras, and massive charnockite and Closepet granite) shows that final mineralogy at magma crystallization depends on local H<sub>2</sub>O activity: charnockite (orthopyroxene-bearing) for low H<sub>2</sub>O activity, granite (biotite-bearing) for high H<sub>2</sub>O activity. The difference in H<sub>2</sub>O activity delimitates two crustal domains: a granulite facies lower crust and amphibolite facies middle crust. The transition between charnockite and granite is progressive, incipient charnockites are situated at the granulite/amphibolite facies boundary.

The H<sub>2</sub>O activity is determined by fluids are found in mineral inclusions. In lower crustal granulites, fluids are

made of dense CO<sub>2</sub> and high salinity brines, buffering the H<sub>2</sub>O activity to ca. 0.2–0.3 at the end of charnockite magma crystallization (F<sub>1</sub>, Fig. 10*b*). The H<sub>2</sub>O activity is much higher (close to 1) in the amphibolite facies middle crust, due to a much greater amount of H<sub>2</sub>O (F<sub>2</sub>, Fig. 10*b*). A great amount of these fluids is locally derived by a variety of processes: progressive dehydration or decarbonation, preferential dissolution of H<sub>2</sub>O in granite melts, oxidation of former organic matter (CO<sub>2</sub>) or, for brines, inherited sediment waters of dissolution of evaporites. These fluids are found in all granulites and granites, but, for UHT granulites, a special characteristic is the great quantity of additional CO<sub>2</sub> (and possibly brines) liberated by the breakdown of carbonatite melts issued from the underlying metasomatized mantle.

Mobile brines can circulate freely in the lower crust, either in the UHT granulite or neighboring normal HT granulite. They can reach the granulite/amphibolite boundary, being responsible for the formation of incipient charnockites. Less mobile CO<sub>2</sub> remains in place as long as the temperature and lithostatic pressure are high enough to maintain a fluid density that is in equilibrium with the surrounding rock.

When the pressure and temperature decreases at the end of the UHT metamorphic episode, CO<sub>2</sub> has to leave the lower crust (Fig. 10c). The CO<sub>2</sub> release is mediated through a system of extension fractures, ranging from crystal-size microfractures to crustal-scale zones, probably related to major earthquakes. Depending on the redox state, CO<sub>2</sub> is either reduced (graphite veins) or oxidized in the form of quartz-carbonate veins (Touret et al., 2019b). Mobile brines can invade pervasively large areas, causing various forms of alkali metasomatism (albitization, scapolitization). The oxygen fugacity is either internally or externally controlled, i.e. through sulfate-bearing brines (oxidized), or sulfide minerals (reduced), respectively.

Noteworthy is the fact that the layer of metasomatized mantle may persist for a very long time at the base of the lithosphere under the stabilized supercontinent (Touret and Huizenga, 2012b; Touret et al., 2016). It reacts to surface rifting by emitting various types of alkaline magmas and carbonatites (Fig. 10d). Examples of these intrusions are best seen on the periphery of the Gondwana supercontinent (Fig. 2), but they do occur in virtually all UHT occurrences worldwide.

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